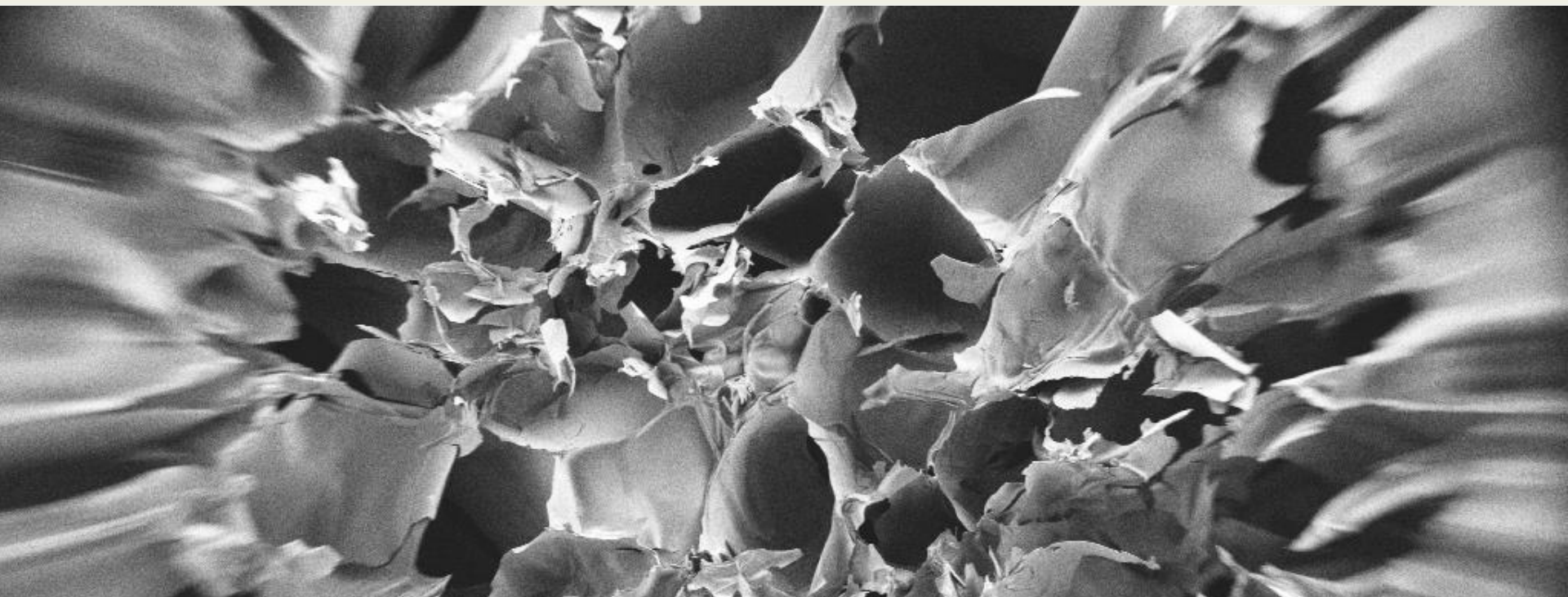




NATIONAL INSTITUTE OF CHEMISTRY



Center for Development,
Demonstration and Training
for Carbon-Free Technologies



Functional Hydrogels for Energy Applications

Superaerophobic Electrode Coatings



Vojtěch Škopek, Filipa A. Andre Vicente, Blaž Likozar, Ilja Gasan Osojnik Črnivec*

gasan.osojnik@ki.si



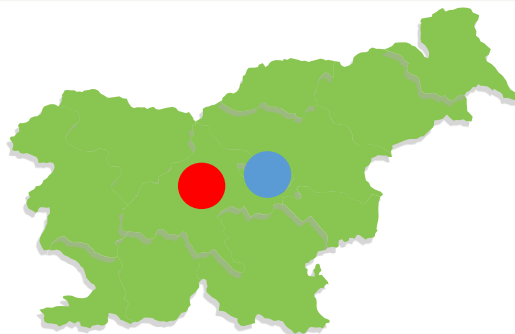
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NATIONAL INSTITUTE OF CHEMISTRY



National Institute of
Chemistry (NIC)
Ljubljana, Slovenia



Center for development,
demonstration and training
for zero carbon
technologies Kisovec,
Slovenia



Undergoing full renovation



Fully operational in 2026, 32 mio EUR investment



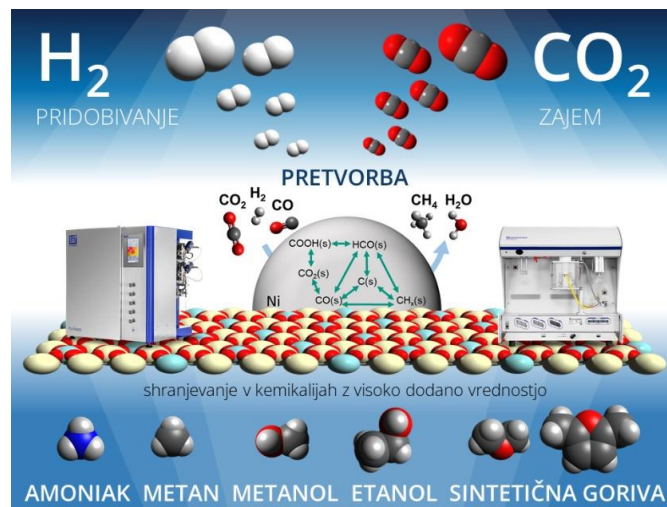
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SLOVENIA



Sofinancira
Evropska unija



L52 Laboratory for Demonstration of H₂ and CO₂ Technologies



Chemical engineering for **alternative energy**, **fine chemical production**, **decarbonisation** and **energy intensive industries**, at industrially relevant operating conditions.

MATERIALS

catalysts, sorbents, electrodes, membranes

CATALYSIS

gas to gas, gas to liquid standard, HT, HP

ELECTROLYSIS

AEM, PEM, water, hybrid

including support to cross-cutting biological processes and waste utilisation challenges



Green Hydrogen

for sustainable industrial processes

ENERGY CARRIER

substitution of fossil energy carriers

SI: glass ✓, ceramic, lime, metallurgic – aluminum, (steel)
long-haul transport: ships, trucks, ...

HEAT RECOVERY

utilization of high temperature industrial heat

ELECTROLYSIS & BIOMETHANE

standalone production of hydrogen

SI: glass and chemical industry

FEEDSTOCK

substitution of fossil hydrogen sources

methanol, ethanol, jet fuel, ammonia etc.



Center for
Carbon-Free



H2GIASS
H₂ furnace



E-CODUCT
H₂ & sulfur
production



Hybrid
SI H₂ and battery
RRI - TRL 3-6

Carbon Capture

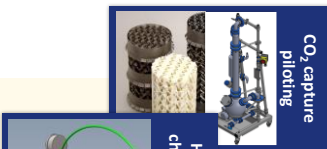
supporting H₂ technologies and biomass recovery

NOVEL SORBENTS & TECH

(CO₂ composition, process heat, gas purity)

FEEDSTOCK

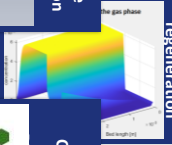
fuels, polymers, advanced ceramics/minerals



CO₂ capture
piloting



High pressure
characterisation



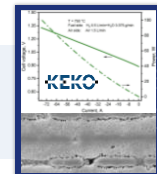
Breakthrough
regeneration



CarbNETS
CO₂ heat &
water

Water Security

supporting decarbonization



SOFC and
H₂ and graywater
treatment

ELECTROLYSIS

Water treatment and alternative water sources

BIOBASED MATERIALS

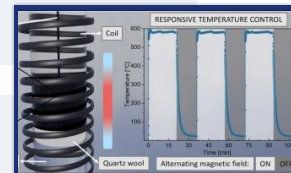
Biobased sorbents for emerging pollutants
and hazardous gasses removal

Biobased PFAS-free advanced coating
and polymer adhesive alternatives

NOVEL PROCESSES

Residual water production

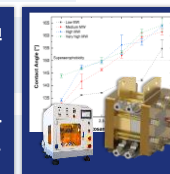
High pressure electrolysis



MICRO
CO₂ to
CH₄ & water



Thermogravimetric
tuning and
sorbent



High P EIS station
membranes and
cell fabrication

Critical Raw Materials

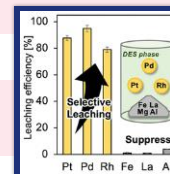
value chains for high added value materials

ADVANCED NANOMATERIALS

novel selective recovery and/or separation of metals

PHOSPHOROUS RECOVERY

novel solvents and electrochemical extraction systems
SI value chain: sludges to technical agents, electrolytes,
electrodes and absorbent producers.



Extraction and
process RRI



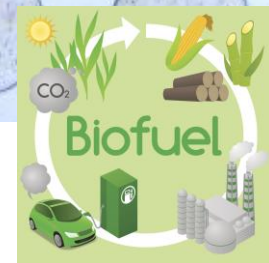
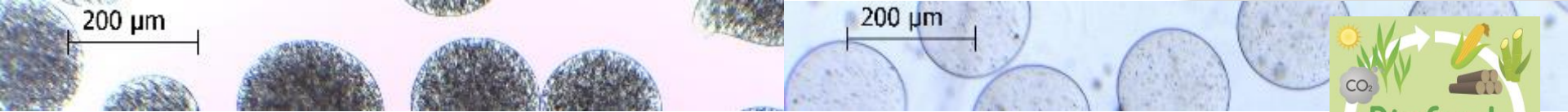
Phosphorous
precipitation and
purification





Features of bio-based hydrogels for protection and delivery





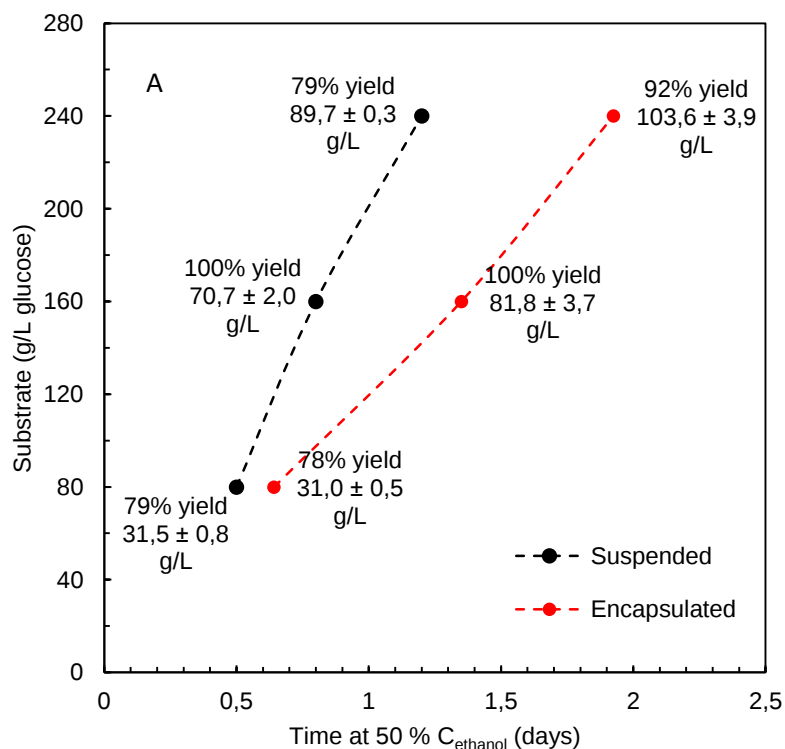
Adverse growth conditions

S. cerevisiae

high gravity conditions

growth medium: YPD @ 8 to 32 wt. % glucose

$V_d = 50$ mL, $T_d = 24$ °C, t_{50} shown on graph



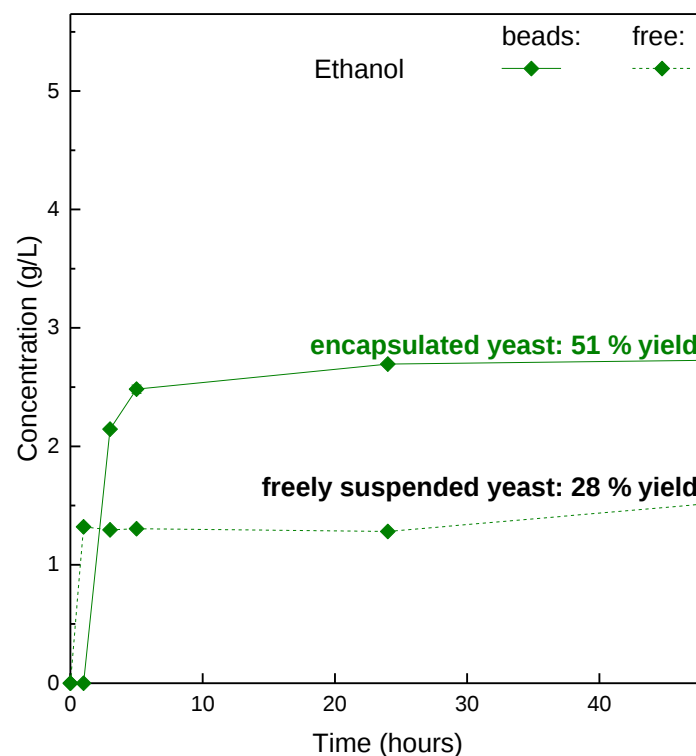
K. Marxianus

presence of organic acids

growth medium: modified MRS

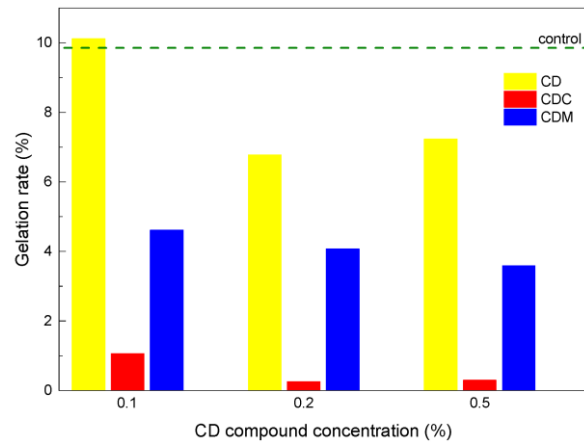
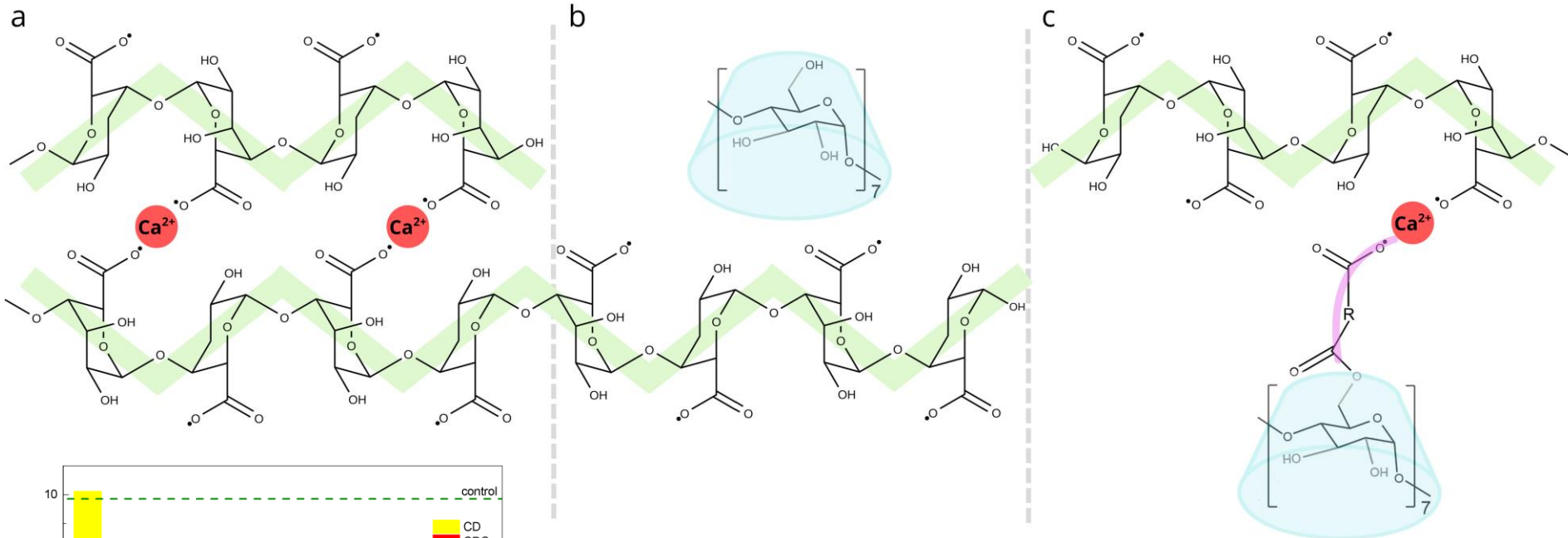
with 1 wt. % lactose, 1 wt. % lactic acid

$V_d = 50$ mL, $T_d = 38$ °C, $t_f = 48$ hours



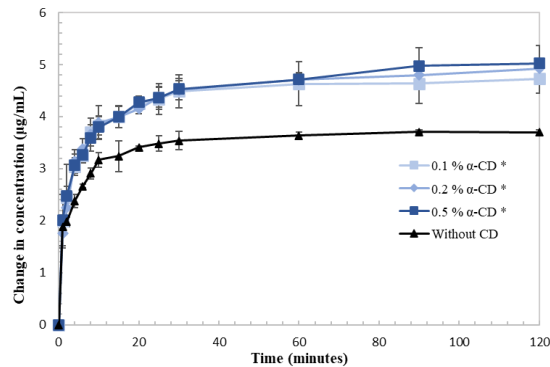
Gel porosity modulation

Alginate gel with
 β -cyclodextrin inclusion



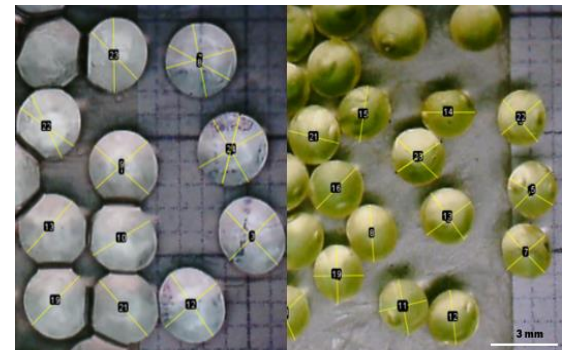
Cyclodextrin reduces solution viscosity (not shown), crosslinked gel deformation resistance (G') and increase flow (G'').

<https://doi.org/10.3390/gels11110875>



Cyclodextrin improves the rate and capacity for hydrophobic compounds integration (example for curcumin).

--in press--

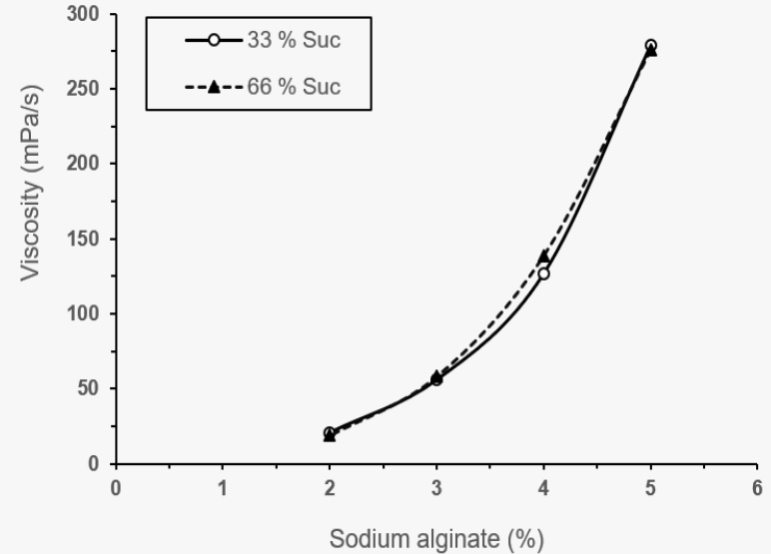


Tuning hydration

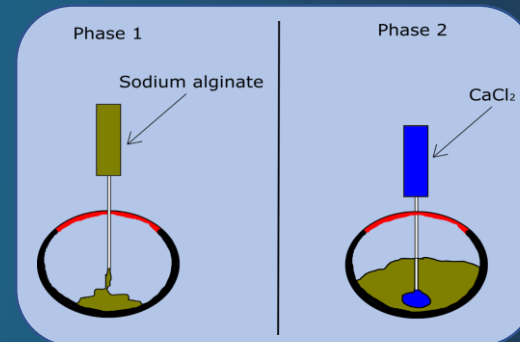


Alginate-based edible inks

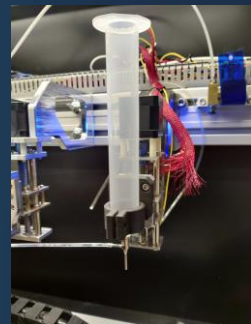
- Good stability in powdered and gel form
 - Good encapsulation properties for gut delivery
 - Vegan acceptable
- Recipe takes in account:
- Fast ion induced gelation (Ca^{2+})
 - Fast solubility/hydration period in cold water (minutes at no/minimal mixing instead hours at intensive agitation)
 - Versatile entrapment (also hydrophobic instead of only hydrophilic)
 - All materials EFSA and FDA approved for food consumption



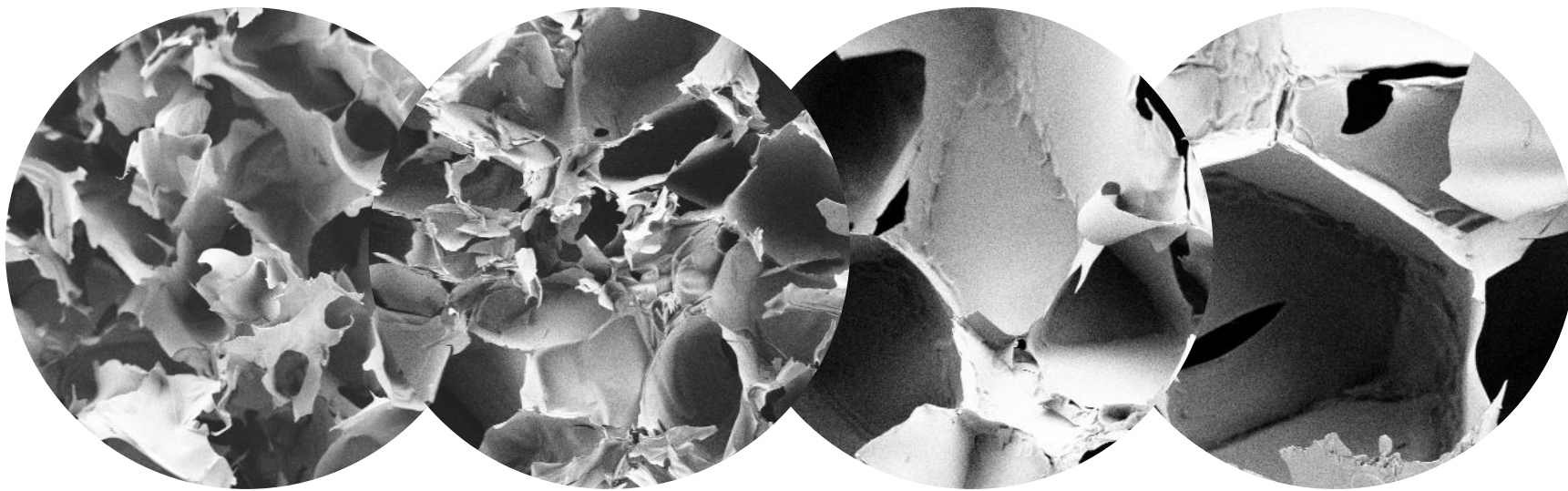
Viscosity of rapidly dissolving alginate matrix



Extrusion steps for the gelation of Na-alginate



Printing nozzle inside the machine



Functional coatings



Hydrogen

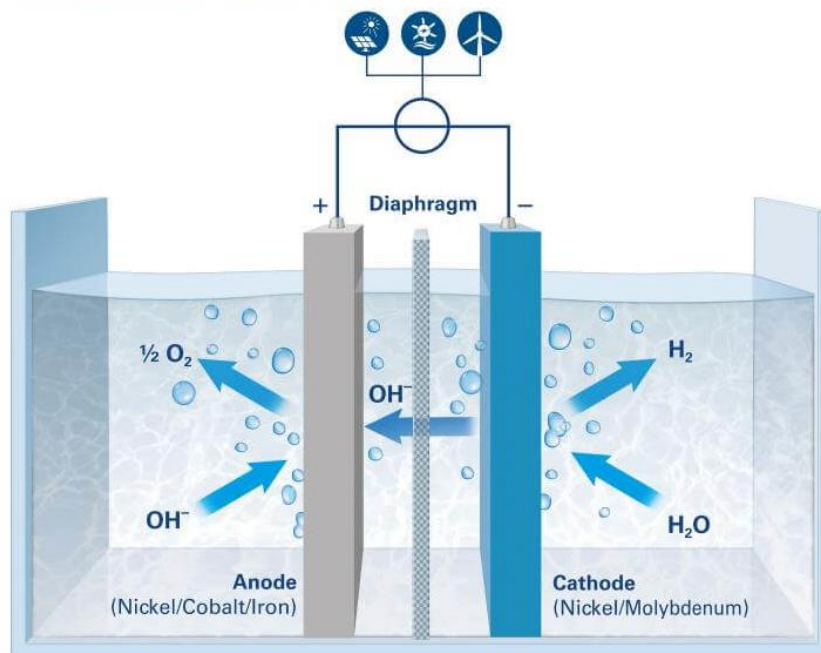
Versatile energy carrier

- the highest **energy content** per mass
- 15x lighter than air
- **Power-X** to for surplus energy

Storage and **transport** challenges, **on demand** production is beneficial.



Alkaline Water Electrolysis

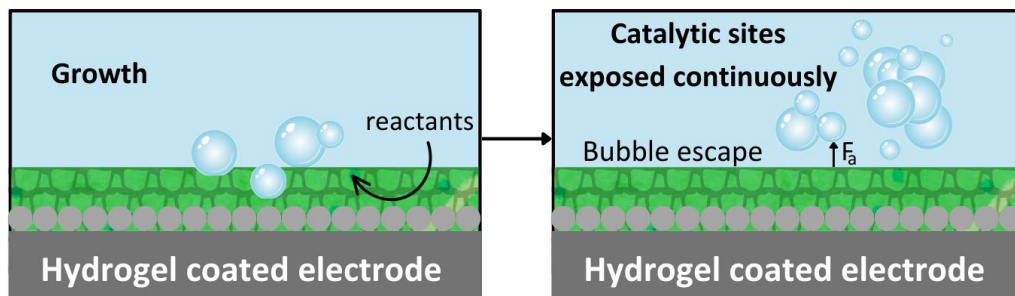
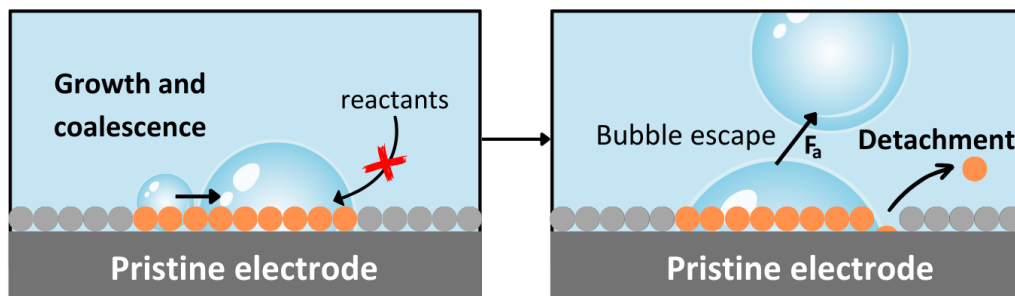
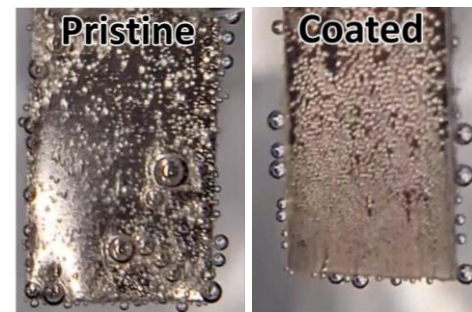


Water electrolysis

- produces 99.99% **pure hydrogen**
- **green hydrogen** with PV/wind
- room temperature and ambient pressure
- Industrial feedstock
- can feed directly into fuel cells

Challenges in hydrogen evolution reaction (HER)

- **high energy consumption**
- **noble metals** as electrodes (e.g., Pt)
- gas bubble at electrodes **reduce efficiency** by blocking active sites
- new alternatives required to boost HER performance



SUPERAEROPHOBIC COATINGS^[1]

- improved bubble detachment
- reduced overpotential
- overall boost in efficiency

However, **toxic** and **expensive**:

- **M13 virus** with specific conditions^[2]
- Fossil based and persistent (polyallylethylamine, polyethyleneimine)^[3]
- toxic crosslinker (glutaraldehyde)

[1] BAE, M. et al. Superaerophobic Polyethyleneimine Hydrogels for Improving Electrochemical Hydrogen Production by Promoting Bubble Detachment. In *Advanced Energy Materials* . 2022. Vol. 12, no. 29, s. 2201452. .

[2] JEON, D. et al. Superaerophobic hydrogels for enhanced electrochemical and photoelectrochemical hydrogen production. In *Science Advances* . 2020. Vol. 6, no. 15, s. eaaz3944. .

[3] KANG, Y. et al. Versatile, Stable, and Scalable Gel-Like Aerophobic Surface System (GLASS) for Hydrogen Production. In *Advanced Functional Materials* . 2024. Vol. 34, no. 2. .



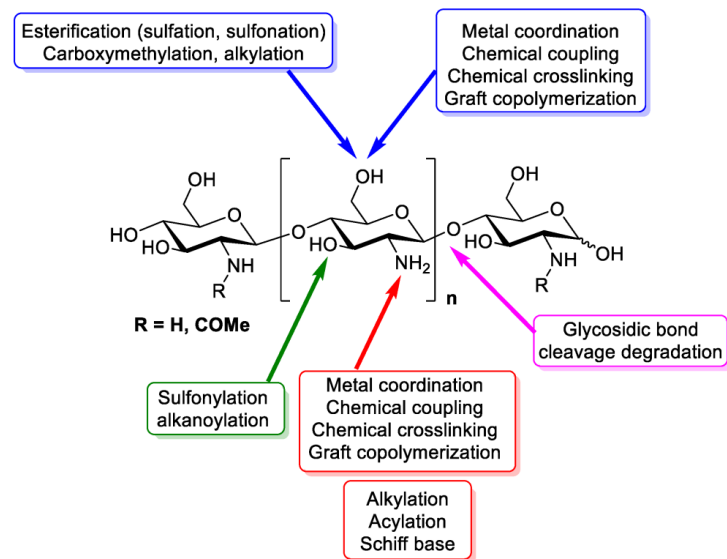
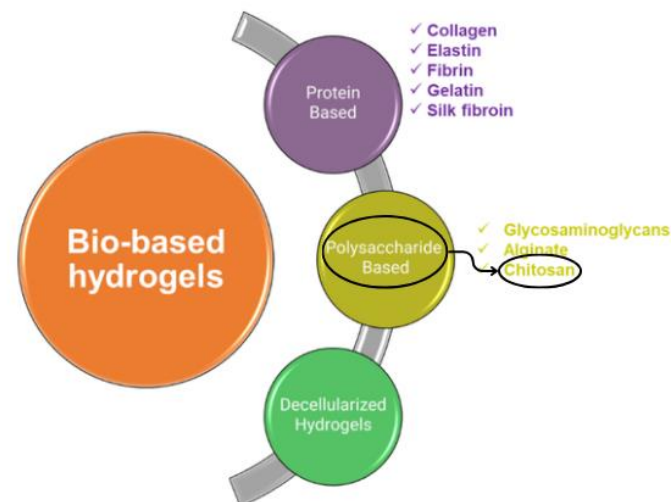
Towards alkaline resistant properties

Chitosan

- Derived from chitin
- Only natural cationic polysaccharide
- Biocompatible, adhesive, modifiable
- High number of hydroxy and amino groups to achieve high hydrophilicity

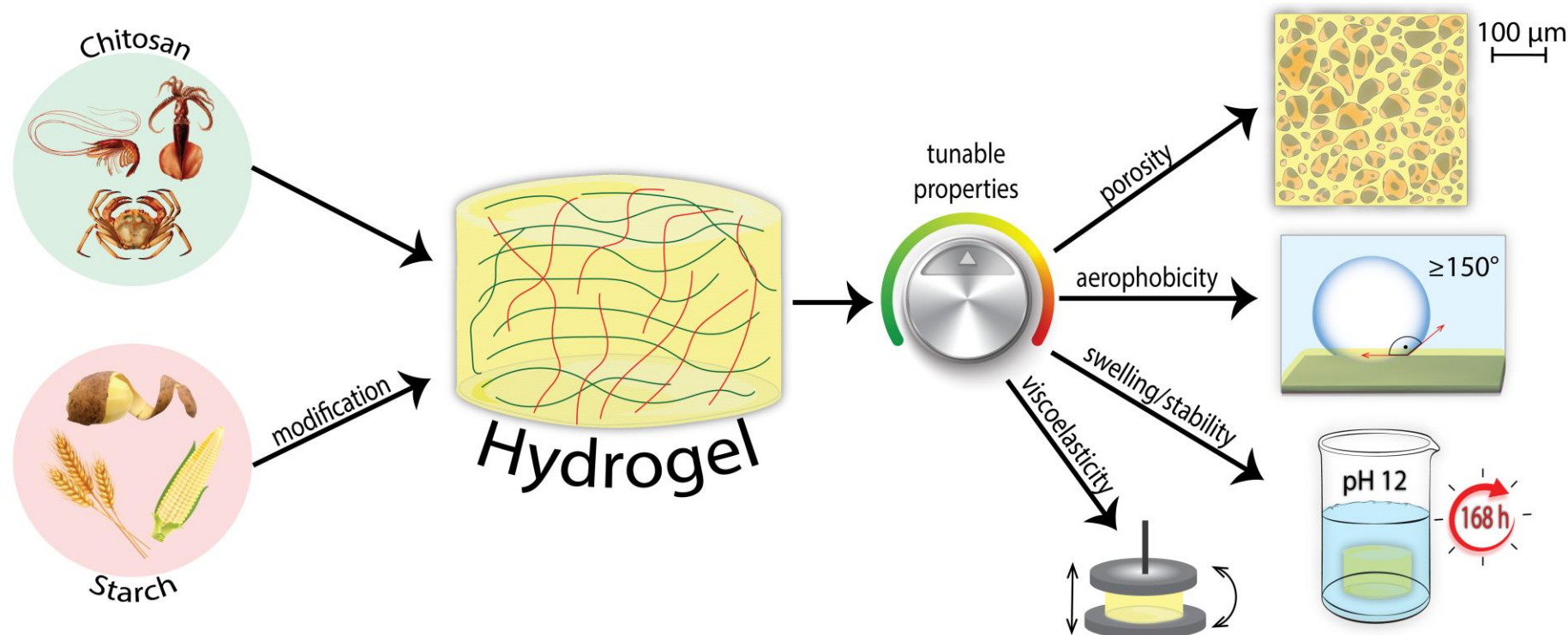
limitations

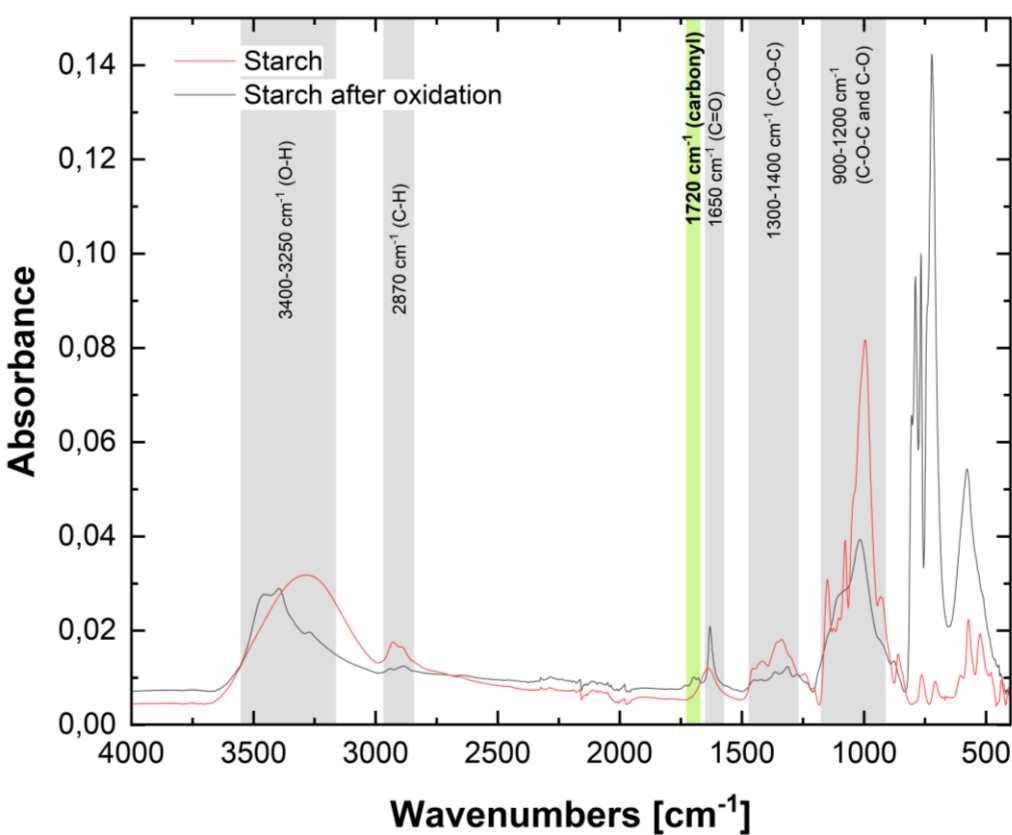
- Common crosslinkers are usually toxic (glutaraldehyde, glyoxal) or expensive (genipin)
- Searching for non-toxic and abundant alternatives





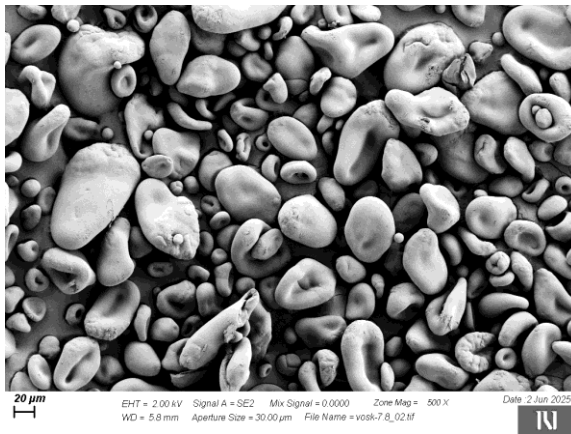
Designing required material properties





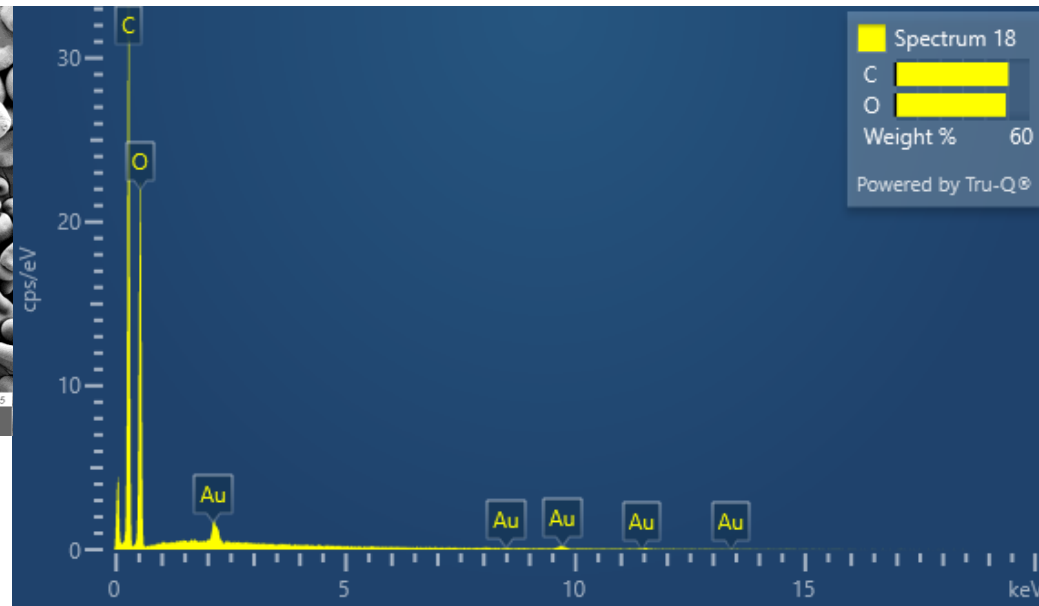
➤ FTIR spectra

➤ Confirming creation of new aldehydic group



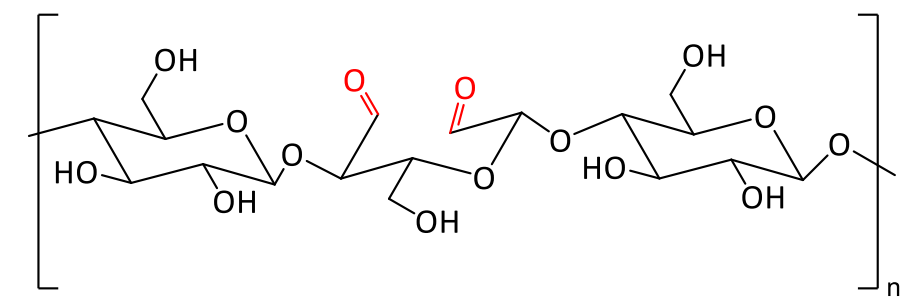
➤ Elemental surface composition

➤ Confirming preparation of pure crosslinker



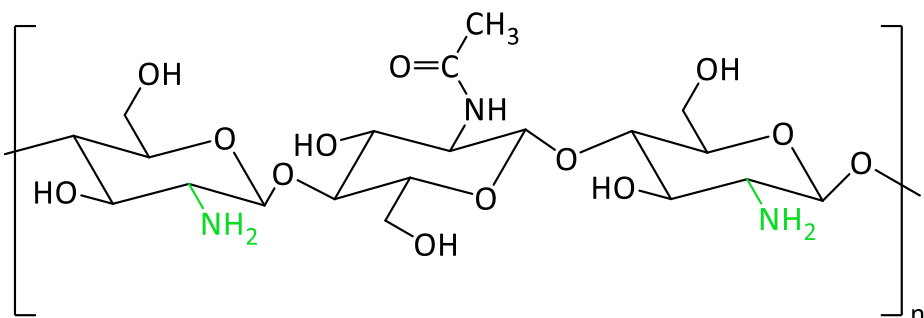


Hydrogel synthesis

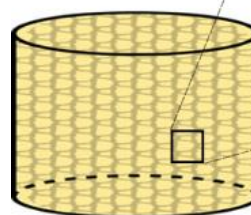


Dialdehydic Starch

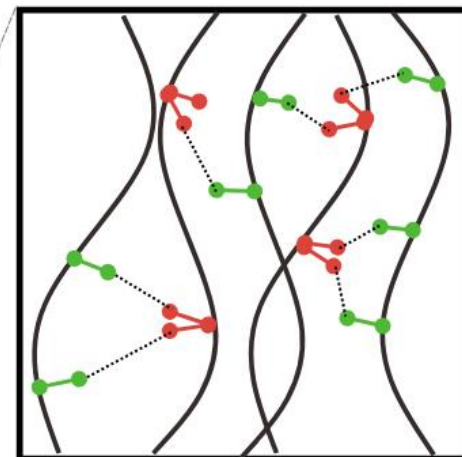
+



Chitosan



HYDROGEL



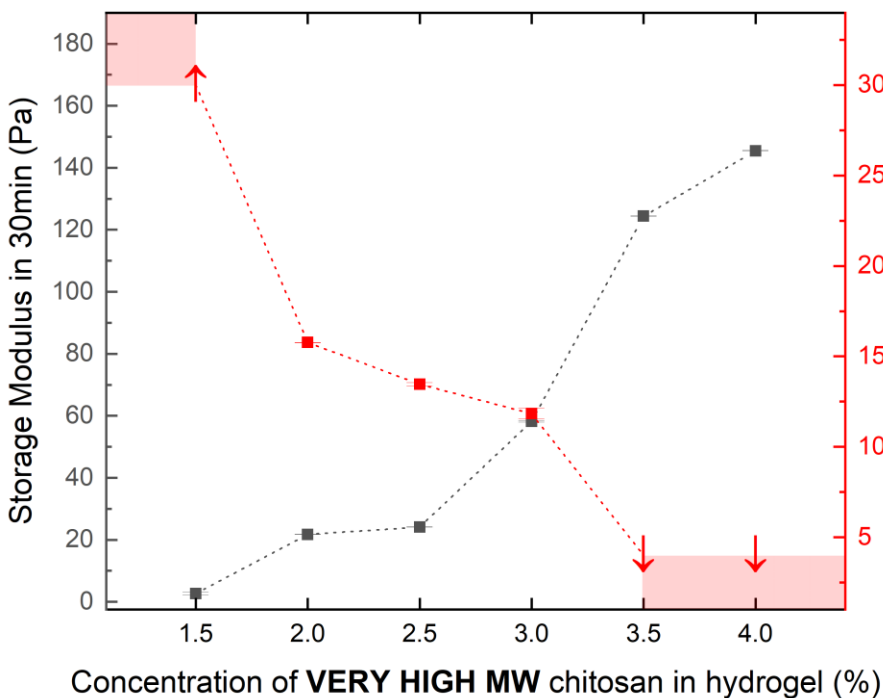
Schiff base reaction
between **amine** and
aldehyde group
forms covalent imine
bond

**to achieve a robust
crosslinked gel**

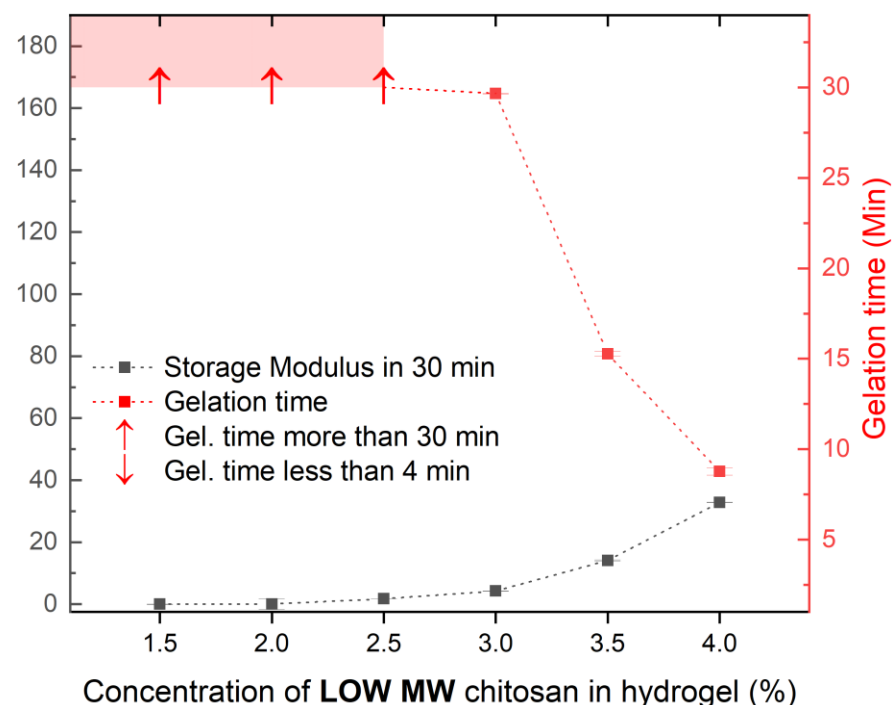


Viscoelasticity and gelation tuning

- Tunability with changing **polymer concentration** and **molecular weight**
- With the right tuning **rapid gelation** (under 4 min)



vs.



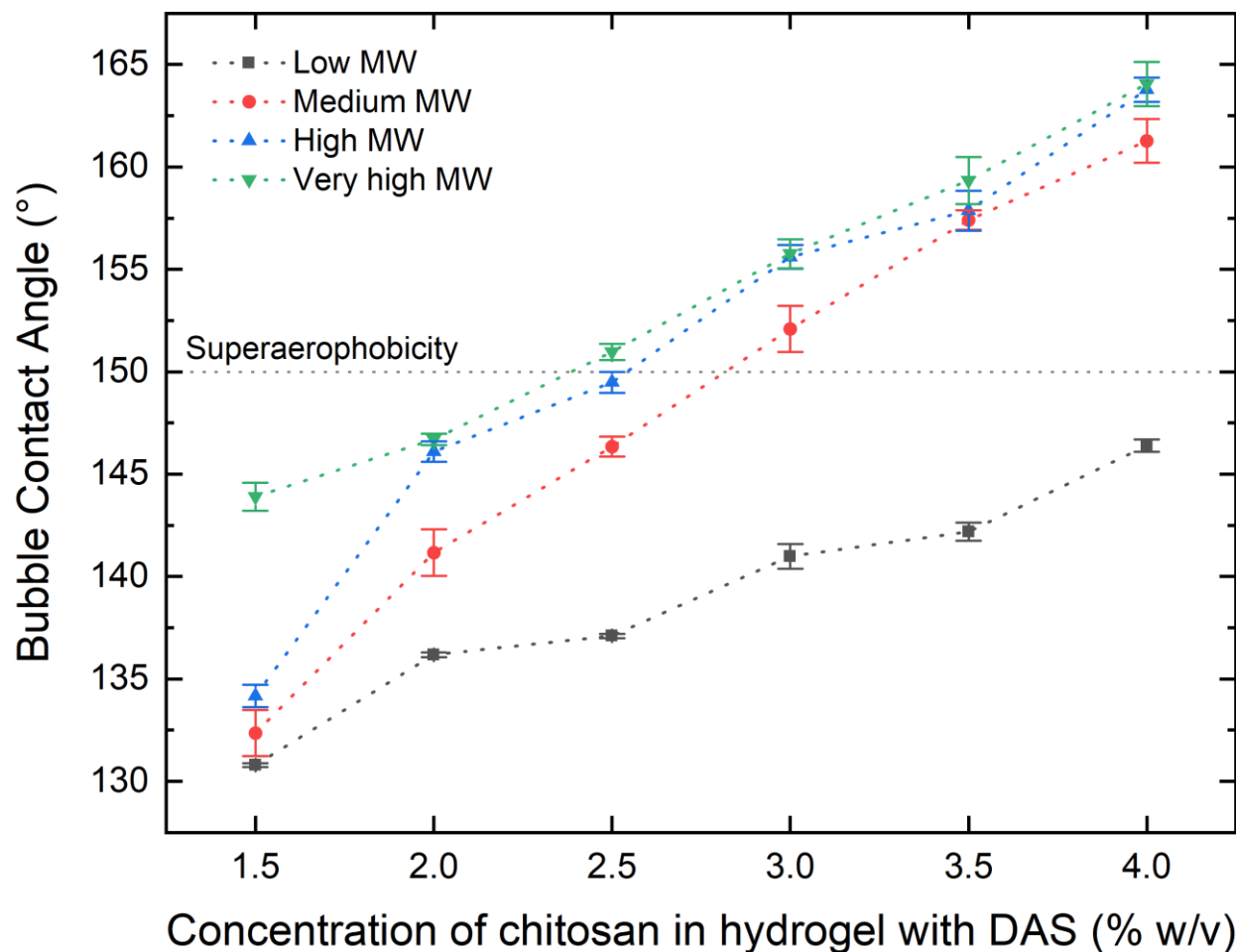
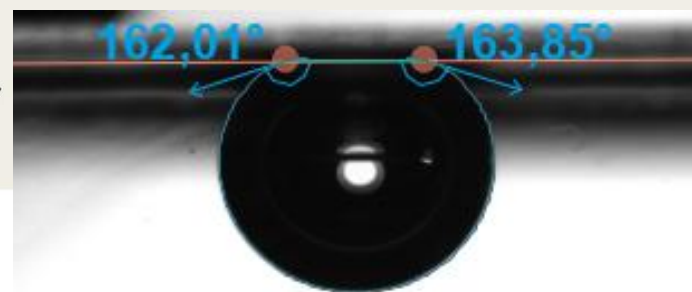


Bubble contact angle

Captive bubble method

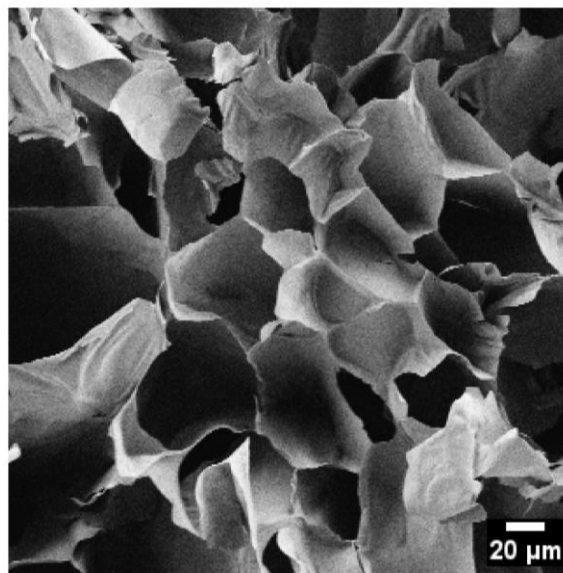
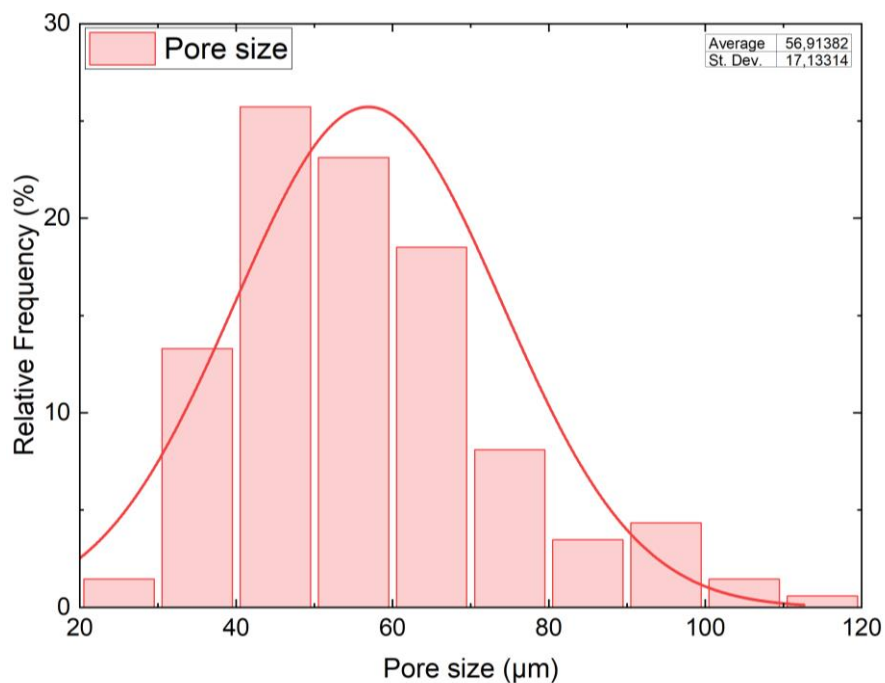
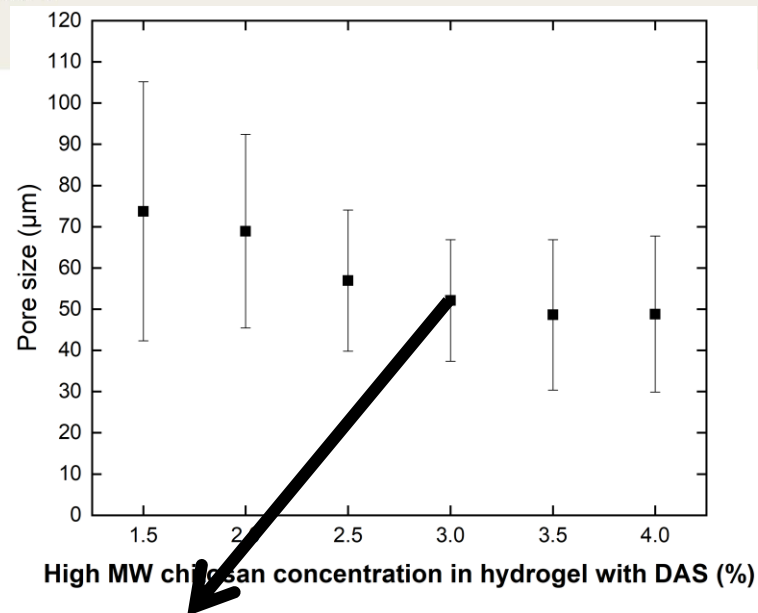
Contact angles
between air bubble
and hydrogel film
under the water

**Superaerophobicity
@150° and above**



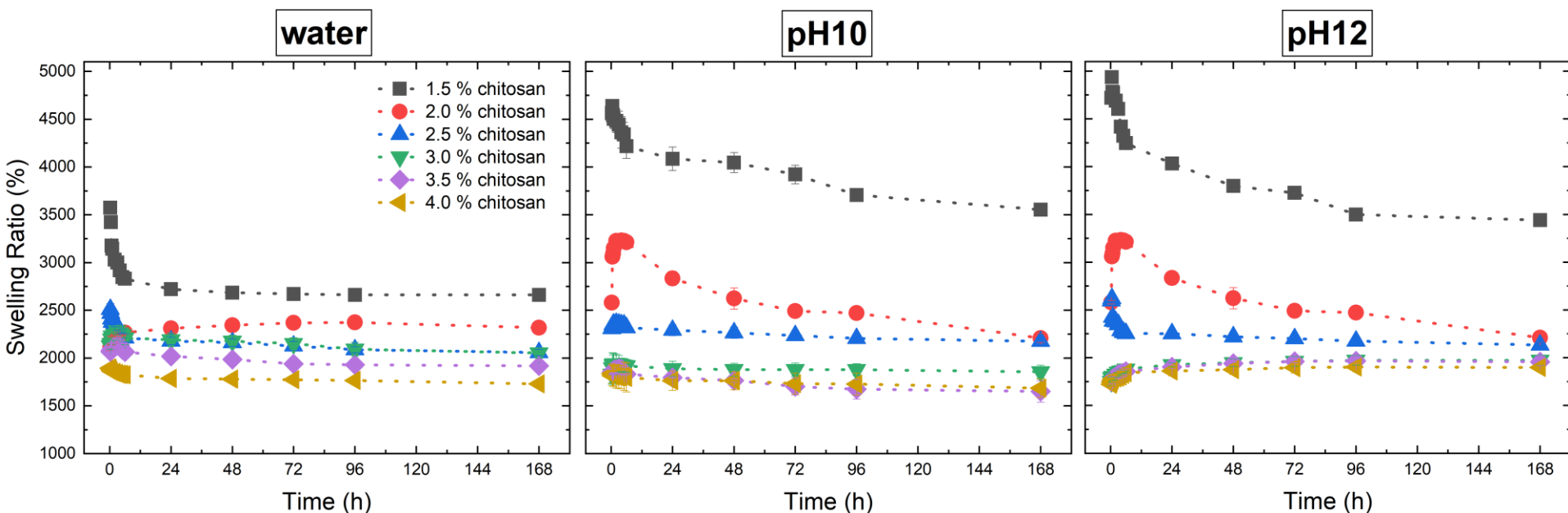
Pore size distribution

- open porous 3D structure
- pore diameters ranging 10-100 μm





Stability in alkaline environment



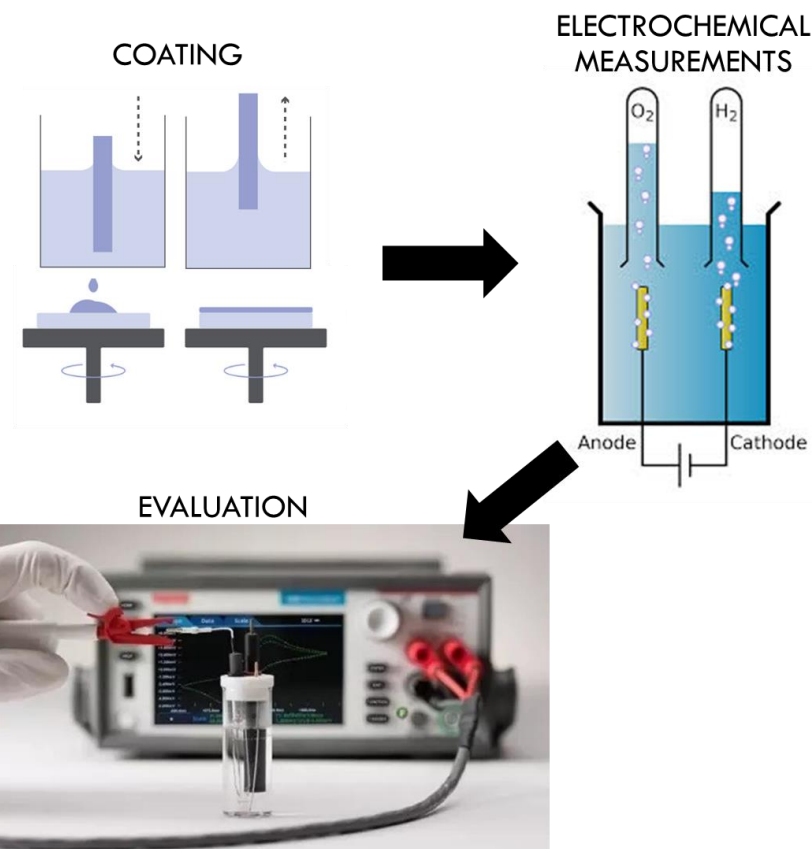
Stable swelling behavior over period of **7 days**



Conclusions and further steps

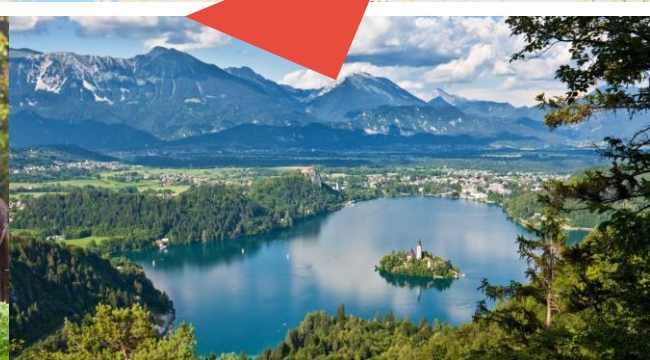
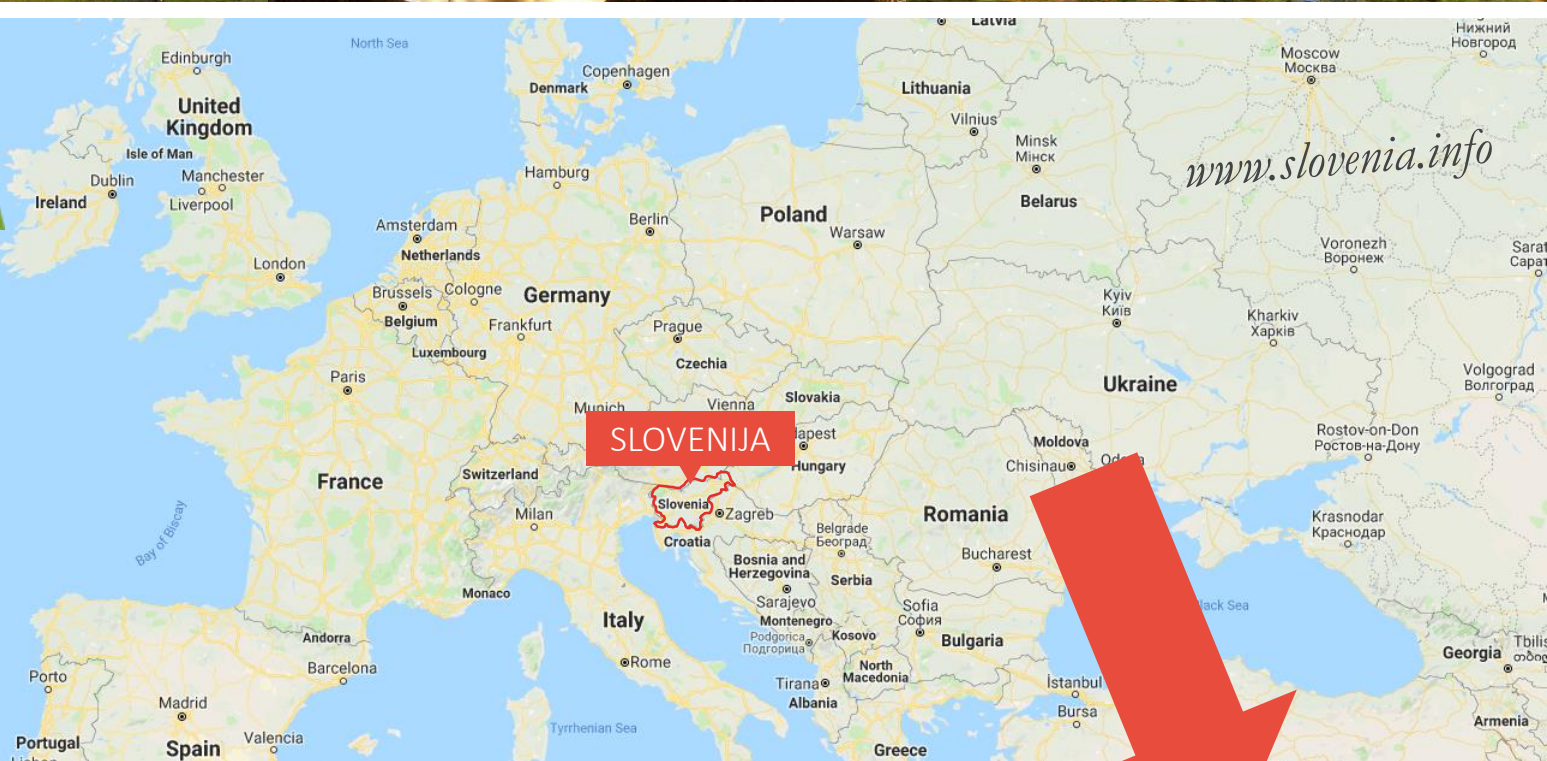
- Hydrogels are customizable well defined chemical environments.
- Shown for **chitosan/DE starch** – **excellent alkaline stability, superaerophobicity** ($\alpha \geq 165^\circ$).
- **Fibers to sustainable hydrogels** to replace **synthetic and PFAS-based ingredients**, particularly for emerging applications of **solid-liquid-gas interfaces** in **harsh alkaline-resistant environments**:

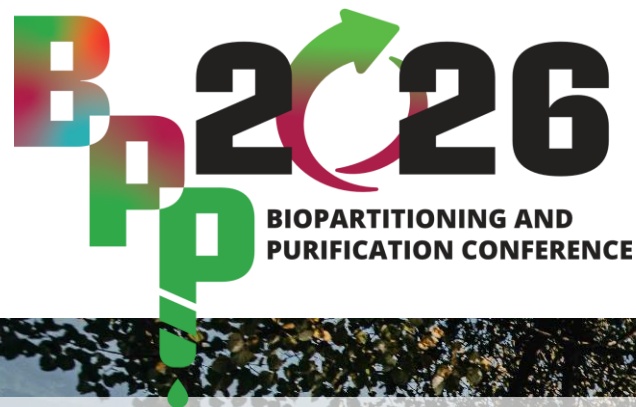
- energy production and storage
- corrosion protective coatings
- environmental absorbents and sensors





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www.bpp2026.si

6 – 9 September 2026

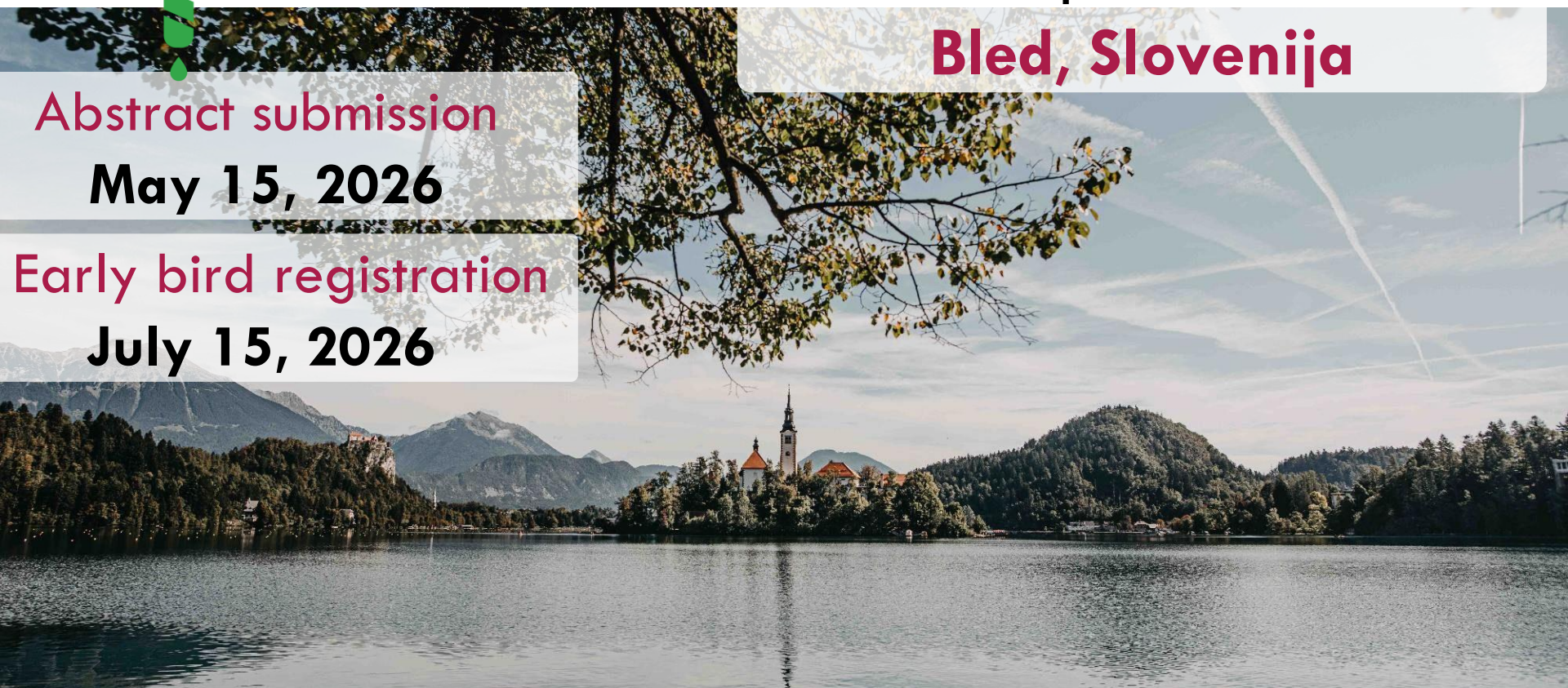
Bled, Slovenija

Abstract submission

May 15, 2026

Early bird registration

July 15, 2026



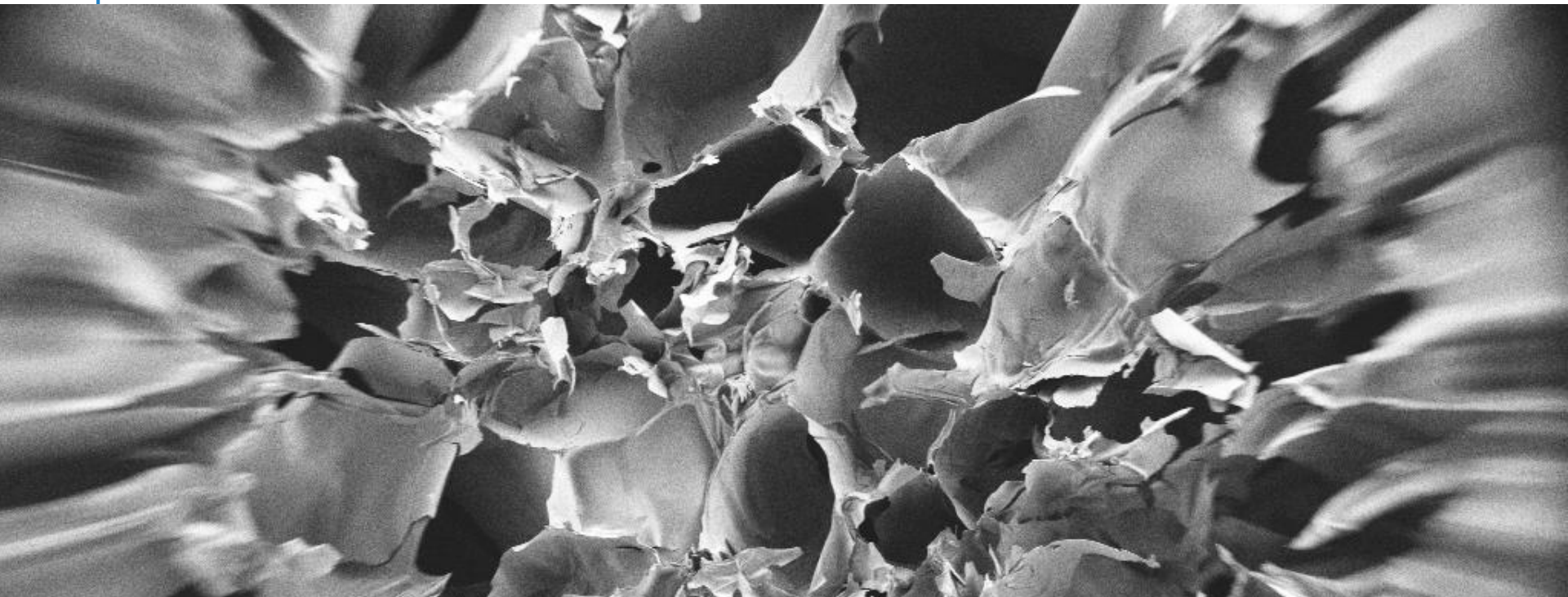
Accompanying workshop: **Fibre101**
10 – 11 September
details follow...



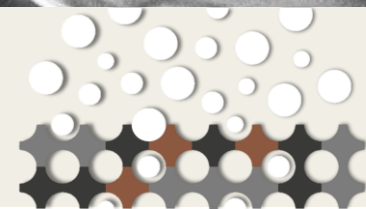
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FUNCTIONAL HYDROGELS FOR ENERGY APPLICATIONS SUPERAEROPHOBIC ELECTRODE COATINGS



Vojtěch Škopek, Filipa A. Andre Vicente, Blaž Likozar, Ilya Gasan Osojnik Črnivec*

gasan.osojnik@ki.si